



# New models for calculating the electrical resistivity of loess affected by moisture content and NaCl concentration

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## Abstract

Saline loess is an important cause of environmental geo-issues in northwest China. In this research, electrical resistivity of loess with different moisture contents and NaCl concentrations was measured at three test frequencies. Results indicates the plastic limit (16%) and around 2% NaCl concentration are the critical content affecting the variation of loess electrical resistivity. The variation and conductivity for conductive paths are affected by moisture content and NaCl concentration respectively. Combined with three-phase composition and diffuse double-layer structure, new models that consider the effect of moisture content and NaCl concentration indicate good applicability in the validation datasets from different types of soil. A new model was verified by comparison between previous studies and experimental results. This research provides (i) theoretical support for the calibration of large-scale electrical field surveys and the observation of saline loess and (ii) a valuable reference for the prevention of environmental geo-crisis and the utilization of soil resources in northwestern China.

**Keywords** Loess · Electrical resistivity · Three-phase structure · Conductive path · Diffuse double layer · Theoretical model

## Introduction

Loess is associated with various environmental geohazards that threaten people and infrastructure in the widespread loess region of northwestern China (Chang et al. 2021a, 2021b; Duan et al. 2021b, 2020; Yin et al. 2021), so it is necessary to investigate the loess properties in this area. The perennial arid environment of northwestern China means that agriculture requires large-area irrigation, which affects groundwater levels and causes serious soil salinization (Weisbrod and Dragila 2006). Previous studies have found that salinity has significant influences on environmental geo-crisis such as landslides, debris flows, and ground collapses (Fan et al. 2017; Fu et al. 2019; Li et al. 2020a; Wang et al. 2019). Therefore, it is urgent to investigate the properties of loess under the influence of water and salt.

In recent years, field surveys using electrical resistance technology (ERT) have been widely used in the investigation of geo-materials, and play an important role in field monitoring and assessment of water infiltration in soil (De Jong et al. 2020; Sendrós et al. 2020; Zhao et al. 2020), groundwater potential assessments (Doetsch et al. 2012; Hasan et al. 2019; Joel et al. 2019), soil liquefaction (Ronald and Ronald 1982), and the stability evaluation of geo-materials (Kemna et al. 2002; Park et al. 2017; Snapp et al. 2017; Zhou and

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Che 2020; Zhou et al. 2002). As a non-invasive method, ERT avoids significant damage to geotechnical materials and has the advantages of wide detection range and simple operation (Tabbagh et al. 2007). However, ERT is often limited by the complex topography and poor working environments of field sites, which can cause inaccurate results (Hen Jones et al. 2014). Meanwhile, the electrical resistivity theory of the Archie model originated from pure sandstone has been extended to a variety of other geo-materials (Cai et al. 2017; Ellis et al. 2010; Rhoades and Schilfgaard 1976). Combined with laboratorial experimental methods, some researchers revealed the electrical resistivity of geotechnical materials from a new perspective (Erzin et al. 2010; Ozcep et al. 2010; Yin et al. 2021). Laboratory electrical resistivity tests have been widely used in the study of silty clay (Rinaldi Victor and Cuestas German 2002), marine clay (Zhang et al. 2014), expansive soil (Chu et al. 2018), compacted kaolin (Cardoso and Dias 2017), and loess (Liu et al. 2014; Zha et al. 2010) for their rapid, economical, and easy-to-use techniques, and accurate test results (Samouëlian et al. 2005). Therefore, it is a feasible method to obtain the characteristic parameters and their variation law of geotechnical materials through laboratory test method (Tang et al. 2018; Zhou et al. 2015), which can provide reference for field geophysical test and reduce the field test error (Islam et al. 2012).

In addition, the theoretical study of electrical resistivity has also attracted the attention of many researchers. Archie (1942) proposed an electrical resistivity model for cohesionless soil and pure sandstone, but it was assumed in the model that solid particles do not conduct electricity. Later, Waxman and Smits (1968), Rhoades and Schilfgaard (1976), and Fortier et al. (2008) improved Archie's model from different perspectives and expanded its application scope. Based on the difference of three-phase conductivity of geotechnical materials (Fukue et al. 1999; Hasan et al. 2018; Yang et al. 2020), some researchers have theoretically deduced the conductivity model of geotechnical materials from the perspective of series and parallel connections (Fukue et al. 1999; Hasan et al. 2018; Tang et al. 2018). Shan et al. (2015) established a new model to characterize the electrical resistivity of frozen soils based on laboratorial tests and combined with the soil structure, and the new model was tested by the validation datasets from the four conditions (water, temperature, density, salt). Nevertheless, most of these studies involve empirical models and few theoretical models of electrical resistivity in loess have been proposed. Furthermore, there has been little research on electrical resistivity in the loess of northwest China.

The aims of this study were twofold: (i) to investigate the effects of moisture content and salt concentration on the electrical resistivity of loess from northwestern China and (ii) to propose a new loess electrical resistivity theoretical model comprised of simple parameters that consider the

influences of moisture content and salt concentration. This result will (i) help to calibrate large-scale field-based electrical-resistivity datasets, (ii) provide a valuable reference for the prevention and control of environmental geo-crisis induced by groundwater changes in loess regions, (iii) obtain a basic data support to explore the environmental status of saline soils, and (iv) allow better utilization of soil resources in northwest China.

## Materials and method

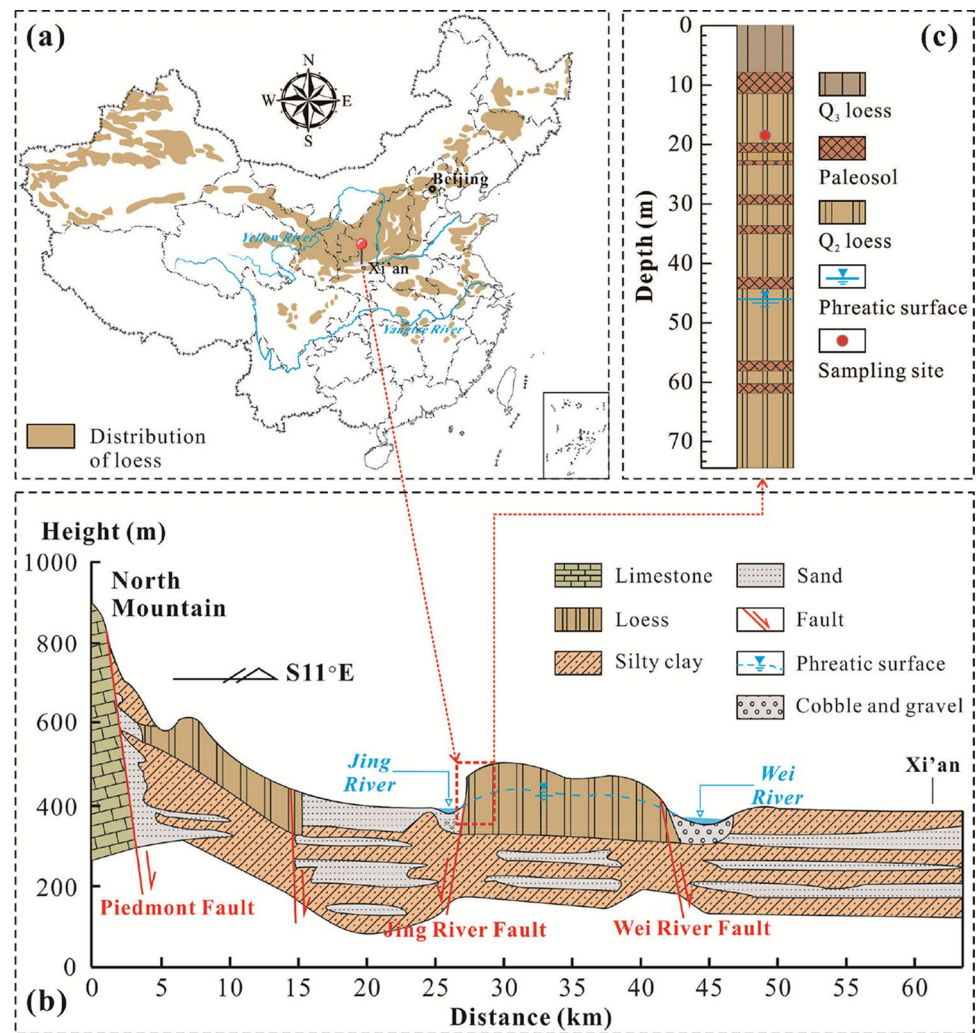
### Materials

Materials tested in this research were Q<sub>2</sub> loess, which were collected from the south Jingyang platform on the southern edge of the Chinese Loess plateau (Fig. 1). Over 62 severe loess landslides occurred in this area since agricultural irrigation began in 1976 (Duan et al. 2021a, 2019; Ma et al. 2021). The loess samples in field were obtained about 17 m away from the ground surface. Its basic physical properties were determined following relevant standard methods (ASTM 2009), and results are presented in Table 1. Particle-size distribution of experimental loess in Fig. 2 was obtained using a laser particle size tester, which indicates that the loess is classified as a silty clay (ASTM-D2487 2017). The mineralogical composition and major chemical composition of loess were determined by an XRD and XRF respectively, and the results are listed in Tables 2 and 3. In addition, Table 3 represents that soluble salt of loess in this area is mainly NaCl, which is also widely mentioned in others' study (Duan et al. 2021b; Yan et al. 2021).

### Sample preparation

The preparation of loess samples consisted of the following five steps: (1) the fresh loess was oven-dried at 110°C for 12 h, subsequently grounded by a rubber hammer, and sieved through 2-mm sieve (Chen et al. 2020; Hu et al. 2021; Li et al. 2021a; Xu et al. 2021b); the above operations can remove organic matter and residual moisture; (2) dried loess was mixed with distilled water and sodium chloride to reach the target moisture content and salt concentration from Table 4, and the mass of distilled water, sodium chloride, and dried loess was calculated using the weighting method; (3) loess, distilled water, and NaCl were mixed by stirring for 10 min with the agitator and sealed in a double-layer plastic bag for 24 h on average 20°C to homogenize the water (Lyu et al. 2020; Xue et al. 2021); (4) static compaction method can control the distance from hydraulic jack, and ensure each sample has the same volume, compactness, and density (Geng and Sun 2018; Melo et al. 2021). Then, the mixed

**Fig. 1** Details of the study area and sampling site. (a) Distribution of Chinese Loess Plateau. (b) Environmental geo-condition in sampling site. (c) Loess profile at the sampling site



**Table 1** Basic physical properties of the loess used in the experiment

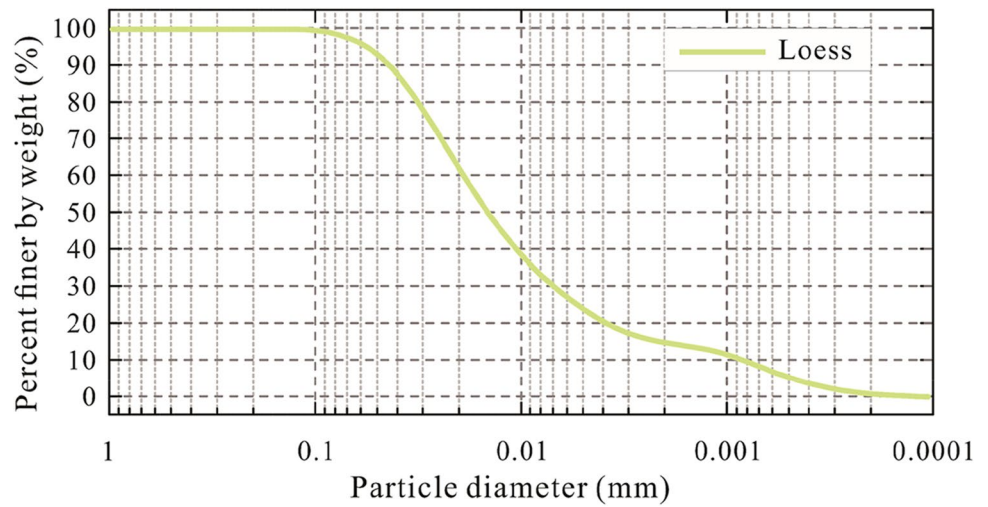
Sample	Moisture content (%)	Dry density (g·cm <sup>-3</sup> )	Void ratio	Plastic limit (%)	Liquid limit (%)	Specific gravity
Loess	7.5	1.44	0.72	16.0	32.1	2.71
Standard deviation	0.10	0.20	0.02	0.5	0.5	0.3

loess was compacted in three layers by a hydraulic jack in the rigid mold (Islam et al. 2012; Wang et al. 2020). Finally, required loess sample was fabricated into a cylindrical specimen with a dry density of 1.70 g·cm<sup>-3</sup> and dimensions of 39.1 mm × 30.0 mm; (5) the error of the sample size, moisture content, and NaCl concentration of all remold loess specimens should be controlled within 0.2% (Liu et al. 2021; Xu et al. 2021a). All specimens were placed in a constant temperature chamber before the test starts. The entire sample preparation procedure was carried out at room temperature to eliminate the influences of temperature on the experimental results.

## Test methods

Existing measurement systems have certain disadvantages: the four-electrode resistivity measurement system disturbs soil structure (Shan et al. 2015), and the DC (direct current) method can change in moisture, pore-fluid chemistry, and soil structure that cause data scatter (Abu-Hassanein Zeyad et al. 1996). Therefore, the LCR digital bridge tester using two-electrode resistivity measurement system and AC (alternating current) was used for this test (Fig. 3, Table 5). Two copper electrodes with a diameter of 40 mm and thickness of 2 mm were placed at the top and bottom

**Fig. 2** Particle-size distribution curve for the loess used in the study



**Table 2** Mineralogical composition of the loess used in the experiment

Item	Quartz	Carbonate	Illite	Kaolinite	Chlorite	Smectite	Others
Content (%)	28	23	21	3	16	6	3

**Table 3** Major chemical composition of the loess used in the experiment

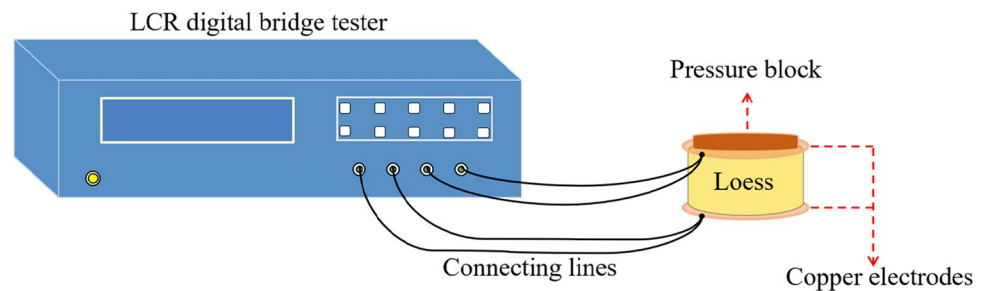
Item	Na <sup>+</sup>	K <sup>+</sup>	Mg <sup>2+</sup>	Ca <sup>2+</sup>	Cl <sup>-</sup>	SO <sub>4</sub> <sup>2-</sup>
Content (g/kg)	42.42	0.58	2.39	12.82	40.26	14.61

**Table 4** Moisture content and NaCl concentration of each sample in the test design scheme

Sample	1	2	3	4	5	6	7	8	9	10	11	12
Moisture content (%)	8	8	8	8	10	10	10	10	12	12	12	12
NaCl concentration (%)	0	2	4	6	0	2	4	6	0	2	4	6
Sample	13	14	15	16	17	18	19	20	21	22	23	24
Moisture content (%)	14	14	14	14	16	16	16	16	18	18	18	18
NaCl concentration (%)	0	2	4	6	0	2	4	6	0	2	4	6

Moisture content (*w*): the ratio of distilled water weight to weight of the dry loess; NaCl concentration (*c*): the ratio of sodium chloride weight to weight of the distilled water

**Fig. 3** LCR digital bridge instrument



**Table 5** Equipment parameters of LCR digital bridge tester

Items	Parameters
Indices	L-Q, C-D, R-Q,  Z -Q
Basic accuracy	0.2%
Frequency range (Hz)	100–10,000
Equivalent circuit	Serial and parallel connection
Output impedance ( $\Omega$ )	30, 100
Trigger mode	Inside
Test signal level (Vrms)	0.3, 1
Measuring range	$R$ (0.1 m $\Omega$ –99.99 M $\Omega$ ); $C$ (0.01 pF–99.99 $\mu$ F); $L$ (0.01 $\mu$ H–99.99 $\mu$ H); $Q$ (0.0001–9999); $D$ (0.0001–9.999)

of a cylindrical loess sample (Liu et al. 2008; Zhang et al. 2014) and connected to an experimental instrument. A pressure block was placed on the top of loess sample to ensure a well contact between the copper electrodes and loess samples (Muñoz-Castelblanco et al. 2012), which can effectively reduce the error (Muñoz-Castelblanco et al. 2012; Shah and Singh 2004). Before the test, open and circuit corrections were applied before conducting experiments and the authors calibrated the LCR tester according to the method of Shah and Singh (2004) and Tang et al. (2018). To eliminate the adverse effect of high frequencies and low frequencies on the electrical resistivity, three test frequencies (100 Hz, 1 kHz, 10 kHz) were used in this research (Duan et al. 2021b). Each loess sample was measured four times at each test frequency, and test results were averaged to minimize the test error. All of the measurements of the resistance were performed under the controlled temperature ( $20 \pm 1^\circ\text{C}$ ). The electrical resistivity of loess is calculated based on the following equation:

$$\rho = R \cdot \frac{S}{L} \quad (1)$$

In which  $\rho$  = loess electrical resistivity ( $\Omega \cdot \text{m}$ );  $R$  = loess resistance measured by measuring ( $\Omega$ );  $S$  = cross-section area through which electrical current conducts ( $\text{m}^2$ ).  $L$  = height of the loess sample (m).

## Results

### Variation of loess electrical resistivity with moisture content

Figure 4 shows the variation of loess electrical resistivity with moisture content under the three test frequencies. It indicates that the reduction effect of moisture content on loess electrical resistivity plays a key role: with the increase of moisture content, the electrical resistivity values of loess

decrease and finally tend to a minimum value. Loess electrical resistivity continues to decrease as the testing frequency grows. Furthermore, with increases in loess moisture content, the slope of electrical resistivity curve decreases. The characteristics of these curves indicate that the effect of water on the electrical resistivity of loess may be gradual. A detailed analysis is provided in the “Discussion” section.

### Variation of loess electrical resistivity with NaCl concentration

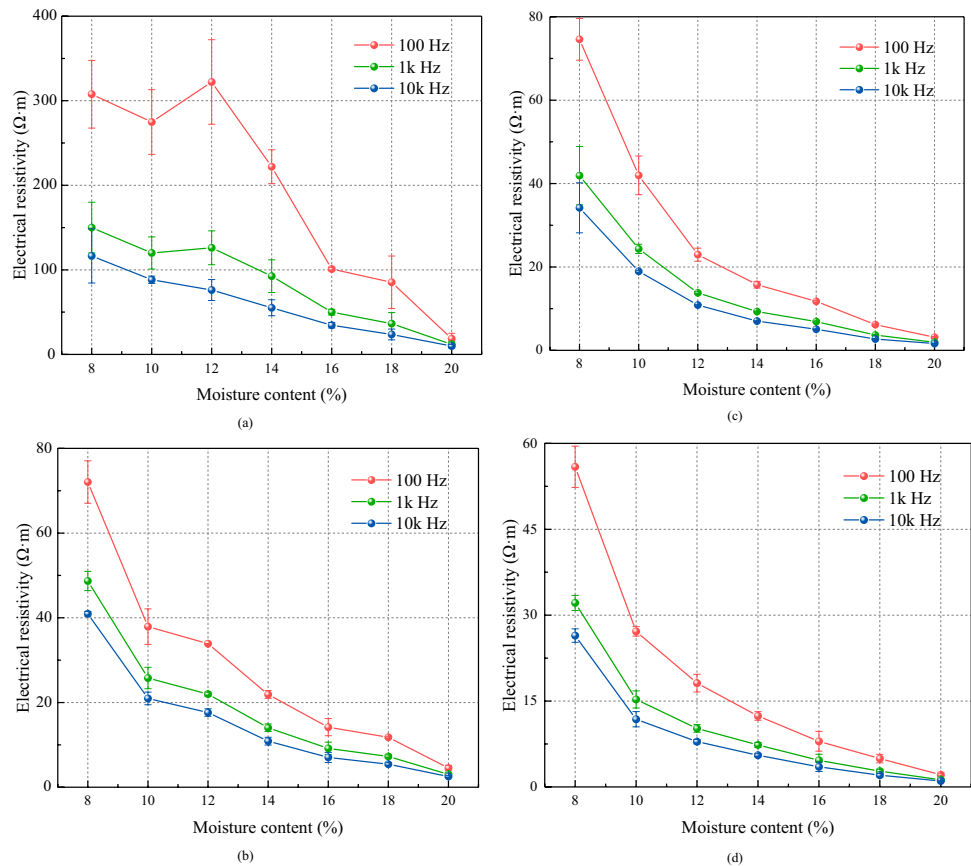
The variation of loess electrical resistivity with different NaCl concentrations under the influence of test frequency is shown in Fig. 5. The effect of salt increases on the electrical resistivity of geo-materials has been mentioned in many studies (Lyu et al. 2019; Rhoades et al. 1989; Zhou et al. 2015). The electrical resistivity of loess is significantly higher at 0% NaCl than at other concentrations. Variations in electrical resistivity characteristics occurred at all moisture contents (from 8–20%). At higher NaCl concentrations of 2–6%, the electrical resistivity decreases overall but at a much lower rate than from 0–2% NaCl. Finally, the minimum electrical resistivity was reached at an NaCl concentration of 6%. The above interesting phenomenon indicates that there is an approximate critical content around 2% NaCl concentration (Duan et al. 2021b; Lyu et al. 2019).

## Discussion

Electrical resistivity models of geo-materials have been systematically studied (Bai et al. 2013; Datsios et al. 2017); however, there are few studies on the loess electrical resistivity. In this study, combined with the three-phase path (Li et al. 2021b, 2020b; Tang et al. 2018) and diffuse double-layer structure (Darrow et al. 2020; Ruedrich et al. 2011), loess conductive models are shown in Fig. 6. Paths 1, 2, and 3 represent three conductive paths: (i) loess particle contact points (Kubliha et al. 2017), (ii) the alternating conductivity



**Fig. 4** Variations in the loess electrical resistivity with moisture content. **a**  $c = 0\%$ ; **b**  $c = 2\%$ ; **c**  $c = 4\%$ ; **d**  $c = 6\%$



of loess particles and pore water (Fukue et al. 1999; Hasan et al. 2018; Mojid et al. 2006), and (iii) continuous free pore water, respectively (You et al. 2020). The physical–chemical interactions between the electrically charged surface of loess particles (mainly clay particles) and the surrounding pore water form a diffuse double-layer structure that surrounds the loess particles, where both anions and cations have conductive capacity under electric field condition rather than insulating (Soga and Mitchell 2005). The above is shown in Figs. 6 and 7.

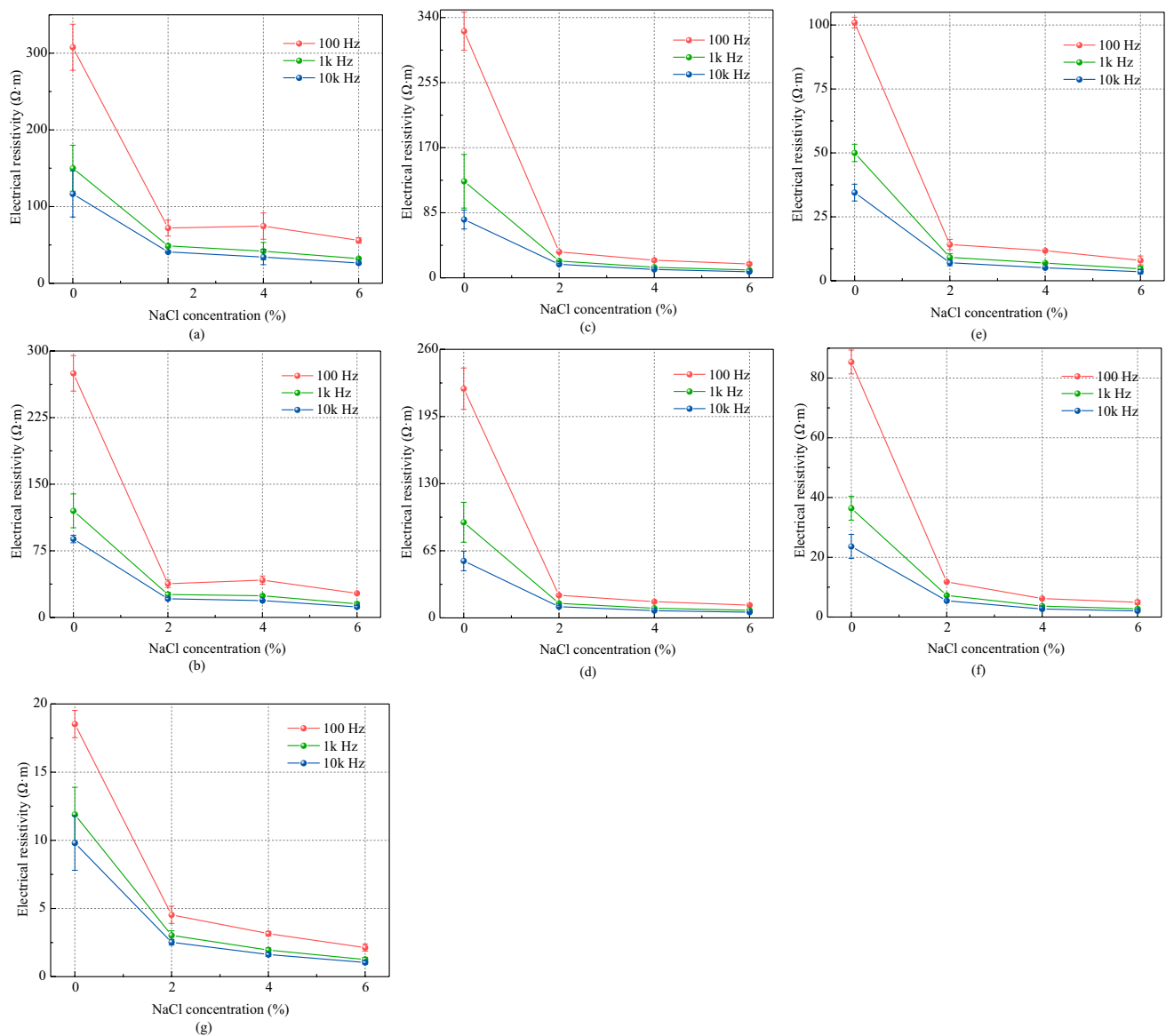
### Influence of moisture content on loess electrical resistivity

At low moisture contents, the conductivity of loess is low, being almost equivalent to that of an insulator (Fukue et al. 1999). Combined with the three-phase path conduction model of loess, variation in the conductive paths in loess in relation to moisture content is shown in Fig. 6. Firstly, due to the lack of pore water, the single point contact mode of loess particles becomes the main conductive path (path 1). In

fact, the electrical conductivity of loess particles is far less than that of pore water (Kubliha et al. 2017; Palacky 1987; Shevnin et al. 2007). Thereby, the electrical resistivity of loess in the initial stage is the greatest and is significantly higher than that at other moisture contents. Secondly, due to the increase of moisture content, some pores in loess are filled with water, the adverse effect of air on the formation of conductive path is reduced, and some path 2 begins to form. Finally, when moisture content exceeds the critical content, the conductive path 3 formed by continuous pore water gradually becomes the main conductive mode. Actually, the plastic limit (16.0%) of soil has been mentioned as a key critical content in many literatures (Fukue et al. 1999; Herring et al. 2019).

### Influence of NaCl concentration on loess electrical resistivity

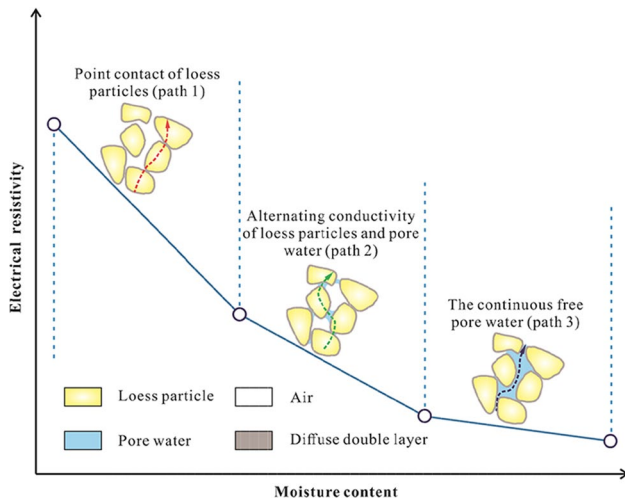
The movement of ions in pore water is the main cause of geomaterial conductivity (Yang et al. 2020). It has been found that as the salt concentration of pore water solution alters, the



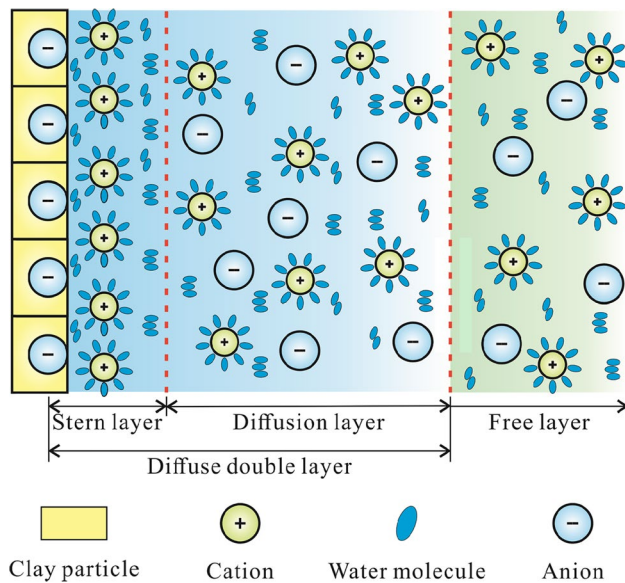
**Fig. 5** Variations in the loess electrical resistivity with NaCl concentration. **a**  $w=8\%$ ; **b**  $w=10\%$ ; **c**  $w=12\%$ ; **d**  $w=14\%$ ; **e**  $w=16\%$ ; **f**  $w=18\%$ ; **g**  $w=20\%$

surface potential of soil particles will alter accordingly, which led to the changes in the thickness of the diffusion double layer of loess (Fig. 7) (Shevnin et al. 2007; Wang et al. 2019). When  $\text{Na}^+$  and  $\text{Cl}^-$  added to the loess, under the influence of the external electric field, the diffuse double-layer properties (i.e., thickness, cation-exchange capacities, and counterion concentration) will change (Shevnin et al. 2007). The relatively free-moving ions from diffusion layer are more likely to produce directional movement to form conductive path under the action

of external electric field, which is also the main reason for the significant decrease of loess electrical resistivity after the increase of salinity. In addition, as the pores of loess are gradually occupied by salt water, some salts and minerals gradually decompose in the pore water, which is beneficial to current transfer (Yang et al. 2020). The conductive path is formed by a large number of free-moving ions in the diffusion layer of clay particles and its surrounding pore water under the action of external electric field. Simultaneously, softening and melting



**Fig. 6** Conceptual diagram of loess conductive path with different moisture content. The red, green and blue arrows represent paths 1, 2, and 3 respectively (Al Rashid et al. 2018; Choo et al. 2016)



**Fig. 7** Structural diagram of the diffuse double layer in loess (Chen et al. 2019; Darrow et al. 2020; Rinaldi Victor and Cuestas German 2002)

of soil particles due to chemical reactions around the pore walls not only expands the pores in the loess, which facilitates the connection of pore water, but also releases free-moving ions that contribute to the conductivity of the conductive path. Thereby, in this stage (*c* is from 2 to 6%), the electrical resistivity of loess decreases continuously.

## Theoretical model and calculations

### Theoretical model

#### Model of loess electrical resistivity under the influence of moisture content

Conduction in loess involves two different processes: (i) pore water in the gaps between loess particles and (ii) the diffuse double layer around the clay surfaces. These processes are closely related to the moisture content, NaCl concentration, and porosity of loess. Figure 8 shows the three-phase conductive system of the loess, where the electrical resistivity is the result of interactions between the pore water, loess particles, and loess structure (Fukue et al. 1999).

Firstly, the following assumptions are made based on the three-phase conductive system of the loess and simplified circuit diagram in Fig. 8.

$$L_s + L_w + L_a + L_1 = 1 \tag{2}$$

$$L_{s1} + L_{w1} = 1 \tag{3}$$

Series and parallel connections are used to characterize the complex connection among three components in soil (Ellis et al. 2010; Mojid et al. 2006). Therefore, the loess electrical resistivity is expressed as:

$$\frac{1}{R_1} + \frac{1}{R_s} + \frac{1}{R_w} + \frac{1}{R_{s1} + R_{w1}} \tag{4}$$

where  $R_1$  represents the loess resistance under the influence of moisture content. According to the electrical resistivity calculation equation ( $R = \rho L /$ ), the following equations are obtained:

$$R_s = \rho_s L_s \tag{5}$$

$$R_w = \rho_w L_w \tag{6}$$

$$R_{s1} = \frac{\rho_s L_1}{L_{s1}} \tag{7}$$

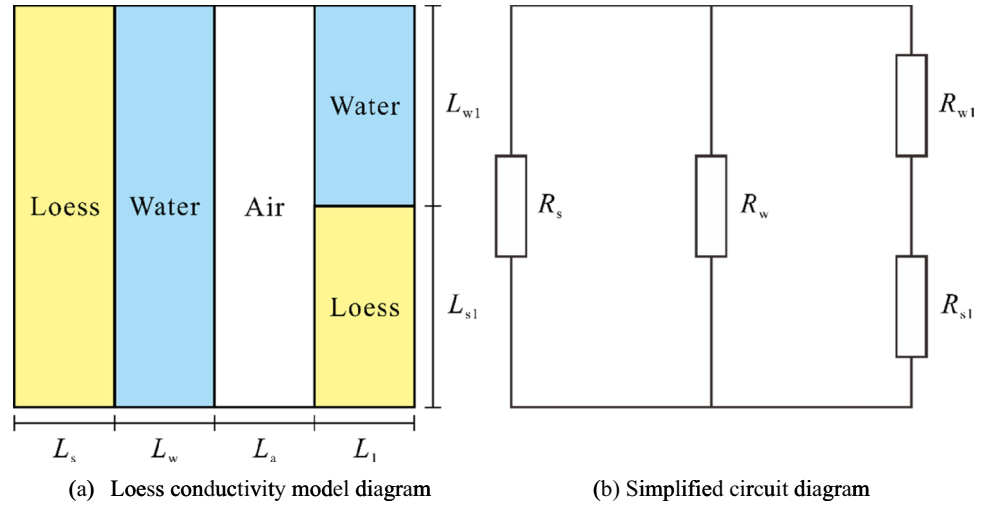
$$R_{w1} = \frac{\rho_w L_1}{L_{w1}} \tag{8}$$

After, the following definitions are given:

$$\alpha_1 = \frac{R_{s1}}{R_s} \tag{9}$$



**Fig. 8** Schematic diagram of the loess conductivity model with moisture content. **a** Loess conductivity model diagram; **b** simplified circuit diagram



$$\beta_1 = \frac{R_{w1}}{R_w} \tag{10}$$

Thereby, the following equation can be obtained:

$$\frac{1}{R_1} = \frac{2\alpha_1 + 2\beta_1 + 1}{\beta_1 \rho_s L_s + \alpha_1 \rho_w L_w} \tag{11}$$

Loess electrical resistivity is closely related to the diffusion double-layer structure around the soil particles. Therefore, the conductivity of diffusion double layer is introduced into Eq. 11 to simplify the model (Waxman and Smits 1968). There is a power function relationship between the electrical resistivity of pore water and moisture content. The simplification process takes into account the diffusion double-layer structure and cation exchange

capacities (CEC). Finally, the electrical resistivity model of loess affected by moisture content is as follows:

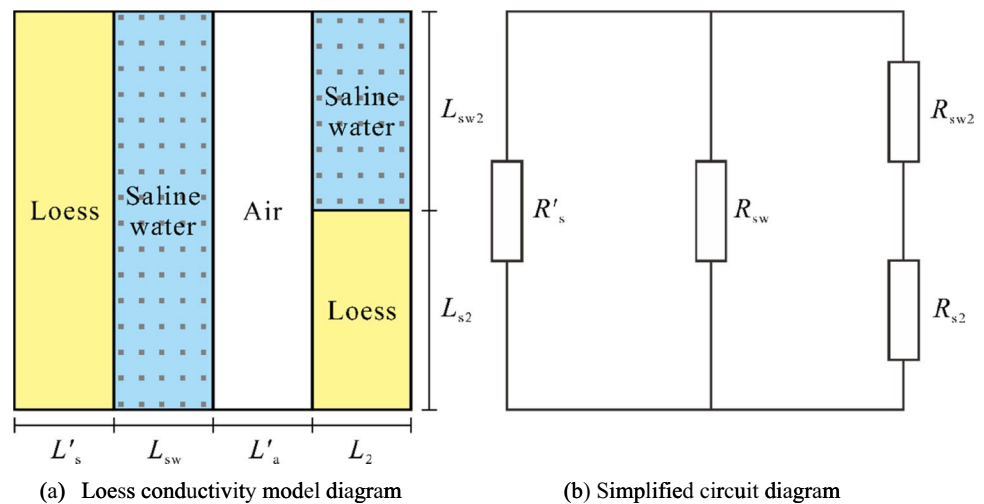
$$\rho_1 = \frac{a_1}{b_1 + c_1 w^\theta} \tag{12}$$

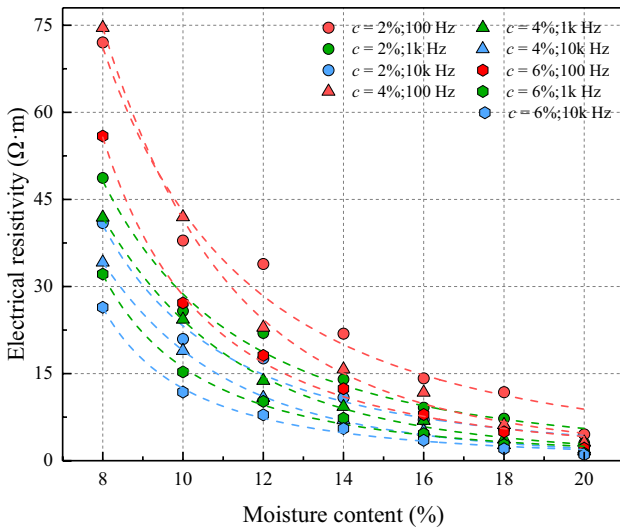
where  $\rho_1$  represents the loess electrical resistivity ( $\Omega \cdot m$ );  $w$  represents the loess moisture content;  $a_1 = 2\alpha_1 + 2\beta_1 + 1$  is the soil structure parameter;  $b_1 = \beta_1 \rho_s L_s$  represents the conductivity parameter of diffuse double layer;  $c_1$  and  $\theta$  represent the resistance parameter under the influence of loess moisture content.

**Loess electrical resistivity model under the influence of NaCl concentration**

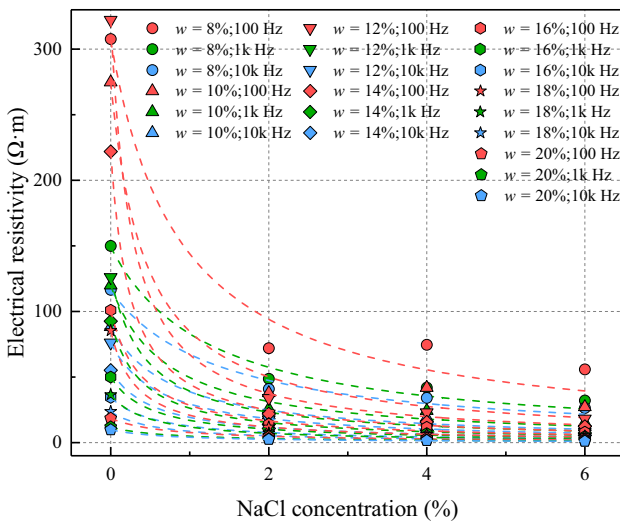
The increased NaCl concentration in pore water allows the entire conductive loess to be regarded as a mixed conductive

**Fig. 9** Schematic diagram of the loess conductivity model with NaCl concentration. **a** Loess conductivity model diagram; **b** simplified circuit diagram





**Fig. 10** Validation results for the electrical resistivity model affected by moisture content based on experimental data



**Fig. 11** Validation results for the electrical resistivity model affected by NaCl concentration based on experimental data

**Table 6** Validation results for electrical resistivity model affected by moisture content based on experimental data

NaCl concentration (%)	Test frequency (Hz)	$a_1$	$b_1$	$c_1$	$\theta$	R-squared values
2	100	1313.500	0.168	0.159	2.281	0.975
2	1 k	814.600	1.234	0.147	2.393	0.984
2	10 k	1090.700	0.930	0.148	2.400	0.981
4	100	4961.100	19.370	0.043	3.369	0.998
4	1 k	2677.900	22.770	0.345	3.400	0.998
4	10 k	19,073.000	161.300	0.350	3.382	0.998
6	100	1843.000	5.755	0.150	2.672	0.995
6	1 k	713.640	6.856	0.150	2.533	0.994
6	10 k	460.950	9.669	0.173	2.431	0.995

body as caused by solid-phase loess particles, liquid-phase saline water, and air-phase void air, as shown in Fig. 9. Therefore, according to the loess electrical resistivity model derivation for the moisture content condition in the “Model of loess electrical resistivity under the influence of moisture content” section, this research provides the following assumptions:

$$L'_s + L_{sw} + L'_a + L_2 = 1 \tag{13}$$

$$L_{s2} + L_{sw2} = 1 \tag{14}$$

$$\frac{1}{R_2} = \frac{1}{R'_s} + \frac{1}{R_{sw}} + \frac{1}{R_{s2} + R_{sw2}} \tag{15}$$

$$R'_s = \rho_s L'_s \tag{16}$$

$$R_{sw} = \rho_{sw} L_{sw} \tag{17}$$

$$R_{sw2} = \frac{\rho_{sw} L_2}{L_{sw2}} \tag{18}$$

where  $R_2$  represents the loess resistance under influence of NaCl concentration.

$$\beta_2 = \frac{R_{sw2}}{R_{sw}} \tag{19}$$

Therefore, the following equation was concluded:

$$\frac{1}{R_2} = \frac{2\alpha_2 + 2\beta_2 + 1}{\beta_2 \rho_s L'_s + \alpha_2 \rho_{sw} L_{sw}} \tag{20}$$

When the loess has a high NaCl concentration, most of the pore water is full of freely moving conductive ions. Pore water can be regarded as a conductive unit closely related to NaCl concentration. Therefore, the equation for loess electrical resistivity is deduced as:

**Table 7** Validation results for electrical resistivity model affected by NaCl concentration based on experimental data

Moisture content (%)	Test frequency (Hz)	$a_2$	$b_2$	$c_2$	$R$ -squared values
8	100	5.866	0.019	0.216	0.974
8	1 k	56.004	0.375	0.300	0.983
8	10 k	64.155	0.554	0.400	0.984
10	100	5.233	0.019	0.042	0.990
10	1 k	3.067	0.026	0.036	0.990
10	10 k	2.624	0.030	0.038	0.990
12	100	4.984	0.015	0.059	0.990
12	1 k	2.502	0.200	0.440	0.990
12	10 k	2.933	0.039	0.061	0.990
14	100	5.340	0.024	0.097	0.990
14	1 k	3.240	0.035	0.088	0.990
14	10 k	3.851	0.070	0.130	0.990
16	100	4.058	0.040	0.103	0.990
16	1 k	3.691	0.074	0.143	0.990
16	10 k	3.311	0.096	0.166	0.990
18	100	3.712	0.043	0.134	0.990
18	1 k	3.342	0.092	0.190	0.990
18	10 k	2.067	0.087	0.152	0.990
20	100	2.466	0.133	0.187	0.990
20	1 k	2.226	0.187	0.262	0.990
20	10 k	1.989	0.203	0.281	0.990

$$\rho_2 = \frac{a_2}{b_2 + c_2c} \quad (21)$$

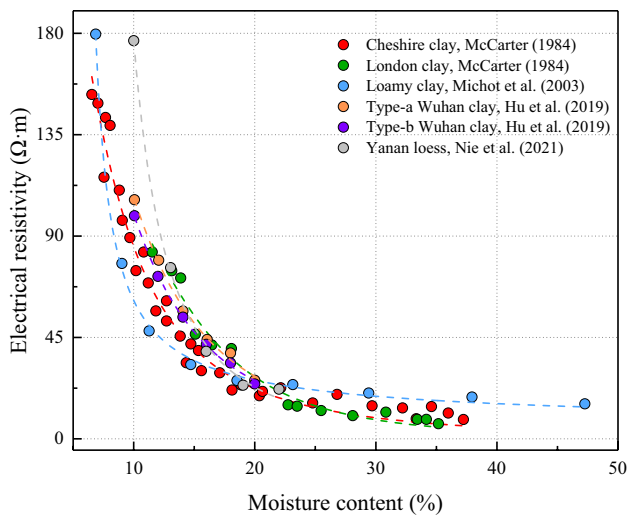
where  $\rho_2$  represents the loess electrical resistivity;  $c$  represents the NaCl concentration;  $a_2 = 2a_2 + 2\beta_2 + 1$  represents the loess structural parameter under the influence of NaCl concentration;  $b_2 = \beta_2\rho_sL'_s$  represents the conductive parameter of the diffuse double layer, and  $c_2$  represents the coefficient under the influence of NaCl concentration.

### Calculation and verification

Data from the loess electrical resistivity experiment were used to evaluate the proposed model. The fitting results under the influence of moisture content and NaCl concentration are shown in Figs. 10 and 11, respectively. Deterministic coefficient ( $R^2$ ) was used to evaluate the loess electrical resistivity model, and the result of model validation is shown in Tables 6 and 7. The determination coefficient represents the difference between the experimental results and the theoretical calculation results, and its value is between 0 and 1. The larger the value is, the stronger the adaptability of the model is (Wang and Hayakawa 1993; Yun et al. 2013). The high degree of similarity between the experimental data and the theoretical data in Figs. 10 and 11 and the expected deterministic coefficient from Tables 6 and 7 show that the

**Table 8** Validation results for new models based on the collected data from different types of soil

Number	Sample name	References	Location	$R$ -squared values	
				Moisture content	NaCl concentration
1	Cheshire clay	McCarter (1984)	Cheshire, UK	0.974	
2	London clay	McCarter (1984)	London, UK	0.969	
3	Loamy clay	Michot et al. (2003)	Paris, France	0.990	
4	Type-a Wuhan clay	Hu et al. (2019)	Wuhan, China	0.990	
5	Type-b Wuhan clay	Hu et al. (2019)	Wuhan, China	0.997	
6	Yanan loess	Nie et al. (2021)	Yanan, China	0.992	
7	Type a-marine clay	Zhang et al. (2014)	Lianyungang, China		0.895
8	Type b-marine clay	Zhang et al. (2014)	Lianyungang, China		0.983
9	Type c-marine clay	Zhang et al. (2014)	Lianyungang, China		0.810
10	Sheyang clay	Zhang et al. (2018)	Sheyang, China		0.934
11	Yancheng clay	Zhang et al. (2018)	Yancheng, China		0.979
12	NaCl-a Yangling loess	Luo et al. (2019)	Yangling China		0.997
13	NaCl-b Yangling loess	Luo et al. (2019)	Yangling China		0.956
14	Na <sub>2</sub> CO <sub>3</sub> -a Yangling loess	Luo et al. (2019)	Yangling China		0.993
15	Na <sub>2</sub> CO <sub>3</sub> -b Yangling loess	Luo et al. (2019)	Yangling China		0.938
16	Na <sub>2</sub> SO <sub>4</sub> -a Yangling loess	Luo et al. (2019)	Yangling China		0.984
17	Na <sub>2</sub> SO <sub>4</sub> -b Yangling loess	Luo et al. (2019)	Yangling China		0.893



**Fig. 12** Validation results for the electrical resistivity model affected by moisture content based on collected data

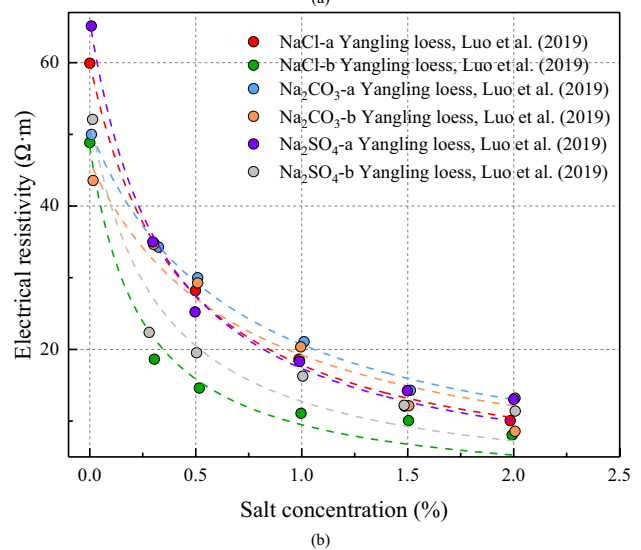
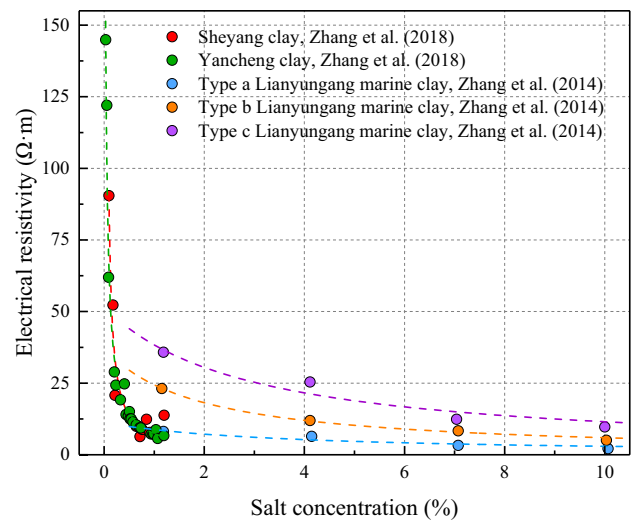
loess electrical resistivity model proposed in this study has good applicability.

In addition, soil electrical resistivity data from different types of researches around the world were collected to evaluate the new models (Table 8; Hu et al. 2019; Luo et al. 2019; McCarter 1984; Michot et al. 2003; Nie et al. 2021; Zhang et al. 2014, 2018). Figures 12 and 13 show the results of fitting the collected data with Eq. 11 and Eq. 21, respectively. The better results between the measured values and the theoretical values, and the high deterministic coefficient, show that the new model has strong applicability.

### Conclusions

The influences of moisture content and NaCl concentration on the electrical resistivity of loess from the south Jingyang Platform, Shaanxi Province, in the Loess Plateau of Northwest China were studied by an LCR digital bridge instrument with three test frequencies. The following conclusions were obtained:

- (1) Test results present the plastic limit (16%) and around 2% NaCl concentration are the critical content of loess electrical resistivity that affected the moisture content and NaCl concentration, respectively. Experimental values gradually decrease and tend to a minimum value after exceeding the critical content.
- (2) The influence of moisture content on the loess electrical resistivity is mainly reflected in the change of the contact model between soil particles and the change of conductive path. As the main factor affecting the



**Fig. 13** Validation results for the electrical resistivity model affected by NaCl concentration based on collected data. **a** Clay. **b** Loess

number of free-moving ions in pore water, NaCl concentration modifies the conductive ability of conductive path.

- (3) Based on the three-phase composition of loess and the diffuse double-layer structure around the solid particles, new models that consider the influence of moisture content and NaCl concentration show good applicability in the experimental data from various countries and regions.
- (4) In addition, the complexity of site conditions requires researchers to conduct more in-depth studies, such as the relationship between the engineering mechanical properties, water sensitivity, thermal conductivity, soil chemical properties, and electrical resistivity.

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**Data availability** All data generated or analyzed during this study are included in this published article.

## Declarations

**Ethics approval and consent to participate** Not applicable.

**Consent for publication** The authors declare that they consent for the publication of this study.

**Competing interests** The authors declare no competing interests.

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